

O₂ variability in Archean environments: Do whiffs trump tiffs?

ARIEL D. ANBAR^{1,2} AND MARK CLAIRE³

¹School of Earth & Space Exploration, Arizona State University, Tempe, AZ 85287 (anbar@asu.edu)

²Department of Chemistry & Biochemistry, Arizona State University, Tempe, AZ 85287

³Department of Astronomy, University of Washington, Seattle, WA 98195 [mclaire@astro.washington.edu]

High-resolution chemostratigraphy in a drill core from the late Archean Mt. McRae Shale, Western Australia, reveals correlated variations in multiple redox-sensitive geochemical tracers, including concentrations of Mo and Re, $\delta^{34}\text{S}$, $\Delta^{33}\text{S}$, $\Delta^{36}\text{S}$, $\delta^{56}\text{Fe}$ and $\delta^{98}\text{Mo}$. These variations are interpreted as reflecting an increase in environmental oxygenation, at least locally, whereby small amounts of O₂ in the water column, and possibly the atmosphere, affected the sulfur cycle and enhanced the rate of oxidative weathering [1, 2].

This “whiff of oxygen” was precisely dated using Re-Os geochronology, obtaining an age of 2500 ± 8 Ma [1]. The timing indicates that the oxygenation episode recorded in the Mt. McRae Shale significantly preceded the Great Oxidation Event at the beginning of the Proterozoic Eon. The accumulated data suggest that typical PO₂ remained $< 10^{-5}$ PAL during this time, and did not persistently cross this threshold until after 2450 Ma.

These findings hint at the possibility that atmospheric O₂ behaved as a biogenic trace gas before the Great Oxidation Event. Results from 1D photochemical models predict short tropospheric O₂ lifetimes (minutes to hours) under a variety of conditions, and the Archean atmosphere may have experienced strong diurnal flux variations because cyanobacterial mats release O₂ during the day and reductants primarily at night [3]. In addition, short O₂ lifetimes would have precluded significant horizontal mixing of O₂, so that Archean ground level O₂ concentrations may have been negligible distant from local “plumes” of biogenic O₂.

The notion of O₂ as a spatially and temporally variable, trace gas in the Archean is, on the one hand, unremarkable. On the other hand, this concept has not been well-integrated into interpretations of Archean redox evolution. Heterogeneity among redox proxies seems likely. As a result, some of the debates that have animated this field over the past decade may prove to have been over false dichotomies.

[1] Anbar *et al.* (2007) *Science* **317**, 1903. [2] Kaufman *et al.* (2007) *Science* **317**, 1900. [3] Hoehler *et al.* (2001) *Nature* **412**, 324.

Redox renaissance

A.D. ANBAR^{1,2} AND G.W. GORDON¹

¹School of Earth & Space Exploration, Arizona State University, Tempe, AZ 85287 (anbar@asu.edu)

²Department of Chemistry & Biochemistry, Arizona State University, Tempe, AZ 85287

The oxygen content of the oceans is intimately connected with the geochemical cycles of carbon and other elements in a complex network of feedbacks. Hence, biogeochemists are interested in reconstructing how redox conditions in the oceans – and in the broader environment – have changed with time. In the wake of recent analytical advances, this decade has seen the invention of several new isotopic paleoredox proxies that may provide information about ocean redox conditions, catalyzing a renaissance of research on this topic. Much of this effort has focused on the Precambrian, an era that witnessed at least two very large, stepwise increases in ocean oxygenation, but these tools are now also being extended into the Phanerozoic. This talk will provide an overview of current research frontiers for Mo and Fe isotopes, two of these new proxies that are now relatively well-established, and an introduction to the insights that may come from U and other complementary isotopic proxies that are still in their infancy.