

Li diffusion and isotopic fractionation in olivines crystals

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In order to constrain the behavior of lithium isotopes in crystals during magma cooling, the study of a fresh pillow-lava with large olivine crystals has been undertaken. An ion-microprobe was used to measure lithium abundances and isotopic compositions both in olivine phenocrysts and in the pillow rim glass. Profiles in $\delta^7\text{Li}$ conducted through olivines show very different patterns correlated with their cooling history. Olivines embedded in the pillow rim glass display no variation in Li abundance ($[\text{Li}] = 1.2 \mu\text{g.g}^{-1}$) and isotopic composition ($\delta^7\text{Li} = +6 \text{‰}$). However, a large Li isotopic zoning is found from crystal cores ($\delta^7\text{Li} = +6 \text{‰}$, $[\text{Li}] = 1.1 \mu\text{g.g}^{-1}$) to rims ($\delta^7\text{Li} = -11 \text{‰}$, $[\text{Li}] = 1.4 \mu\text{g.g}^{-1}$) for olivines set in microcrystalline groundmass. The cores of large olivine phenocrysts are equilibrated with the pillow rim glass ($\delta^7\text{Li} = +6 \text{‰}$, $[\text{Li}] = 3.4 \mu\text{g.g}^{-1}$).

Recently, detailed investigation of the Li isotopic composition of large phenocrysts in various magmatic environments demonstrated that microscale $\delta^7\text{Li}$ variations were due to fractionation of lithium isotopes during late stage chemical diffusion (Barrat *et al.*, 2005; Beck *et al.*, 2006; Jeffcoate *et al.*, 2006). In our case, the diffusion-induced Li isotopes fractionation is the only process able to explain such diverse zoning in olivines from one single pillow lava. Our study brings new constraints on the Li fractionation during the diffusion process and suggests that the pristine $\delta^7\text{Li}$ values can be completely erased in small phenocrysts, even in the case of rapidly cooled volcanics. The use of Li isotopes in crystals for tracking mantle heterogeneities should be taken with extreme caution.

References

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$\delta^7\text{Li}$ systematics of mantle xenoliths from Kilbourne Hole: Unraveling metasomatic & diffusional processes

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We present a detailed Li isotopic study of spinel-peridotites from Kilbourne Hole, New Mexico, including whole rock and hand picked mineral separate $\delta^7\text{Li}$ data (measured by MC-ICP-MS) and in-situ $\delta^7\text{Li}$ data measured by ion microprobe. From this detailed approach the effects of mantle metasomatism and diffusion can be unravelled.

The xenolith suite can be divided into those which are in textural equilibrium, and those which have interacted with the host lava. Whole rock values for the sample suite range from +4.5 ‰ to -2.9 ‰. Of the samples which are in textural equilibrium, mineral separates reveal that olivine, orthopyroxene, and clinopyroxene are all in isotopic equilibrium, and grains analysed by ion microprobe are homogeneous in both Li and $\delta^7\text{Li}$ across grains, with the exception of the outer 50 μm in some clinopyroxene grains. The two lightest $\delta^7\text{Li}$ are found in equilibrated LREE enriched harzburgites, which is consistent with these samples having completely equilibrated with a melt with a slightly negative $\delta^7\text{Li}$ (-3 ‰), possibly from a garnet pyroxenite/eclogite source (although partial equilibration with a heavy $\delta^7\text{Li}$ melt can also explain these data).

Kilbourne Hole peridotite xenoliths have rapid transit times (~0.2 days), which is long enough to produce the Li isotope perturbation observed in the clinopyroxene grains by more rapid diffusion of ^6Li compared to ^7Li from the host lava.

The extent of clinopyroxene reaction correlates with a decrease in whole rock $\delta^7\text{Li}$, which can be explained by reaction driven diffusion with the host lava. Li diffusion in clinopyroxene is at least ten times faster than in olivine [1], and so interactions between peridotite and melt produce clinopyroxene with lighter $\delta^7\text{Li}$ than co-existing mineral phases. Clinopyroxene data that have very light $\delta^7\text{Li}$ values (e.g. [2]) have been used to argue a very light $\delta^7\text{Li}$ component in the mantle. However, without evidence of Li isotopic equilibrium, the existence of such a component must be treated with caution.

References

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