Quantifying mineral dissolution and xenolith digestion rates

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Minerals and xenoliths dissolve in magma as it falls or risesrelative to the magma. Such convective dissolution rates areoften higher than diffusive dissolution rates and can be quantifiedusing the theory Kerr (1995) and Zhang and Xu (2003). In thiswork, I use the theory to carry out a parametric investigation of zircon dissolution rate and xenolith digestion rate. The digestion rate of a xenolith is given as -dr/dt =bD/dwhere -dr/dt is the dissolution rate, D is diffusivity of the maindiffusing component (such as MgO for olivine-rich xenolith), d is he boundary layer melt thickness for a xenolith sinking freelyrelative to the melt, and b is a dimensionless compositional parameter $(C0-C\infty)/(Cx-C0)$ where C is the concentration of themain diffusing component and the unit is mol/L, C0 is the interfacemelt concentration, $C\infty$ is the initial melt concentration, and Cx is the concentration in the xenolith. Given temperature, xenolithcomposition and melt composition, the parameter b can beestimated from thermodynamics of the melt, and diffusivity D canbe found from literature. Hence the calculation of xenolithdigestion rate becomes the problem of calculating the boundarylayer The results show that the boundary thickness d. layerthickness does not change significantly with increasing xenolithsize, and is roughly proportional to the (1/3) power of viscosity.We are constructing a viscosity model that is able to calculateviscosity of all natural melts. The density difference betweenxenolith and melt also affects the boundary layer thickness. For example, at 1300° C, assuming D = 10-11 m2/s, viscosity $\eta = 10$ Pa·s for a Hawaiian basalt, b = 0.08 for olivine and 0.16 forpyroxene, the boundary layer thickness is about 0.072 mm, and the dissolution rate is about 0.04 mm/hr for olivine and 0.08 mm/hrfor pyroxenes. The ascent velocity of magma must be more than 0.19 m/s to entrain a xenolith of radium 50 mm to bring it to thesurface. Zircon dissolution rate and "immortality" of zircon willalso be investigated.

References:

1. Kerr (1995) Contrib. Mineral. Petrol., 121, 237-246.2. Zhang & Xu (2003), Earth Planet. Sci. Lett., 213, 133-148.