Melting equilibria of F-bearing silicic magmas: an experimental study

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Fluorine is an important volatile constituent of granitic magmas, which are often linked to economic ore deposits. Chemical interaction of fluorine with silicate melts creates significant changes in their equilibrium and transport properties (liquidus- and solidus-temperature lowering, viscosity decreases, water solubilities). However, differentiation paths, saturation limits by fluorine-bearing solid phases and immiscible liquids remain unknown. Experiments performed in cold-seal pressure vessels (500-800 °C, 100 MPa) and quenching furnaces (800-1100 °C, 1 atm) along sections in the albite-K-feldspar-quartz- F_2O_{-1} - H_2O system serve as a model for the investigation of the latemagmatic and hydrothermal evolution of F-bearing granitic suites.

At near-solidus conditions (600 °C, \leq 100 MPa), feldspathic assemblages coexist with topaz/corundum, cryolite or villiaumite (in order of increasing alkali/alumina ratio). At low $f_{\rm F20-1}$, topaz or corundum are incompatible with villiaumite and at high $f_{\rm F20-1}$, albite does not coexist with chiolite.

Fluorine interaction with silicate melts results in the formation of Na-Al-F associates and a highly-polymerized SiO₄ framework, which implies large stability of silica polymorphs and leads to fluoride-silicate liquid-liquid immiscibility at supraliquidus conditions. The miscibility gap extends from the cryolite-silica binary (> 1060°C, 1 atm) through peralkaline albite-silica-cryolite ternary (> 930°C, 1 atm) into subaluminous albite-silica-cryolite-topaz quaternary $(> 750^{\circ}C, \le 100 \text{ MPa})$ at anhydrous conditions. The fluoridesilicate miscibility gap rapidly closes at hydrous conditions and common quartz-feldspar-bearing magmas saturate with solid cryolite and/or topaz. Increased quartz stability causes Fbearing silicate magmas to become feldspathic or even feldspathoidal by extensive quartz fractionation. Residual liquids in the haplogranite-F2O-1-H2O and haplotrondhjemite-F₂O₋₁-H₂O systems are SiO₂-poor (~ 53 wt. %), weakly peraluminous (A/NK ~ 1.15), F-rich (13-18 wt. %) and crystallize at 540°C and 580°C (at 100 MPa, 10 wt. % H₂O), respectively.

Stabilities of fluoride solids are determined by activity products of their components. Fluorite saturation severely limits the fluorine enrichment (< 0.2 wt. %) in most Ca-rich metaluminous magmas. In low-Ca melts, topaz stability is controled by interrelated effects of melt peraluminosity and its fluorine content (Al₂O₃^{excess} * F₂O₋₁ ~ 9 mol. % at 720°C, 1 kbar, H₂O-saturation). During degassing of hydrous F-bearing silicate magmas, high alkali-aluminofluoride solubilites and hydroxy-complexing in aqueous vapour suggest the formation of alkaline solute-rich fluids. These can explain contrasting natural occurrences of quenched F-rich topaz ongonites *versus* intrusive F-poor topaz-free feldspathites, which lost their volatiles into hydrothermal aureoles.

Nitrogen 15 signals of anthropogenic nutrient loading in *Anemonia sulcata* from the Adriatic Sea

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Coastal and estuarine environments near urbanized area and tourist centres along the Adriatic coast are more or less subject to pollution due to anthropogenic nitrogen inputs, which can have profound detrimental effects on coastal ecosystems. The objective of this study was to test the hypothesis that the ¹⁵N content of Anemonia sulcata tissue collected near shore and offshore along the eastern coastal part the Central and Northern Adriatic is a reliable indicator of the anthropogenic nitrogen impact on coastal Adriatic water ecosystems. The results shown in Figure 1 indicate that $\delta^{15}N$ of Anemonia sulcata tissue was significantly higher at the polluted sites of Pirovac Bay, Prosika, and the Western Istrian coast than in the reference location in the unpolluted Kornati Islands. The $\delta^{15}N$ enrichment was as high as 5.93‰ and is larger that would be expected from natural variations alone. The δ^{13} C values of sewage-affected Anemonia sulcata also suggests enrichment with light ¹²C isotope, most probably due to a higher degree of anthropogenic nutrition. Our results indicate that $\delta^{15}N$ of Anemonia sulcata tissue appears to be a promising indicator of anthropogenic N loading if caution is used in selecting reference sites.

Figure 1: δ¹⁵N and δ¹³C variations in *Anemonia sulcata* tissue (Central Adriatic: - Kornati Island, - Pirovac Bay, - Prosika, - Pri_njak; Northern Adriatic: - Western Istrian coast)

