# Influence of scale on trace metal dynamics in sediment and water.

## W. DAVISON<sup>1</sup> AND M. MOTELICA-HEINO<sup>1</sup>

<sup>1</sup>Department of Environmental Science, Lancaster University, Lancaster, LAI 4YQ, UK (w.davison@lancaster.ac.uk)

Uptake of trace metals by organisms can affect local chemical concentrations, while the concentration in turn determines the component's availability to organisms. To witness and understand these processes it is necessary to make measurements and develop models on a scale appropriate to the chemical perturbation. In water columns, where transport is rapid, relatively coarsely spaced measurements can show transient features that may only be present for days. In sediments transport is slower and much finer scales are required. The technique of DGT (diffusive gradients in thin films) accumulates metals in situ at a binding agent after their passage through a defined layer of hydrogel. It can make measurements of metals at a spatial resolution of 35µm in two dimensions. In sediments a microstructure of small-scale (100µm to mm) metal remobilisation sites can be measured. The steep, but highly localised concentration gradients of spherical geometry are introduced by local biological activity. The biology is controlling the micro- rather than the macroscale chemistry. We have yet to appreciate fully the extent to which the micro-scale chemistry affects the biology. Combined probes that can simultaneously measure metals and sulfides, show that sulfide and metals can be remobilised at the same location, again suggesting biological origin. The focus on small-scale changes in chemical concentrations is providing new understanding of the in situ mechanisms of interaction between solutes and solid phases in sediments. Understanding the biologically-induced, 3-dimensional chemical gradients on the micron scale is providing a new paradigm for appreciating and quantitatively modelling the biogeochemistry of sediments.

# The new wave in geochemistry publishing - electronic journals

S. J. R.  $DAY^1$  AND S. A.  $WOOD^2$ 

 <sup>1</sup> Royal Society of Chemistry, Thomas Graham House, Science Park, Cambridge, UK (geochem@rsc.org)
<sup>2</sup> Department of Geological Sciences, University of Idaho,

Moscow, Idaho, USA (swood@iron.mines.uidaho.edu)

## **Geochemical Transactions**

The launch of *Geochemical Transactions* in 2000 saw a collaboration between two of the World's most respected chemical societies, the Division of Geochemistry of the American Chemical Society and the Royal Society of Chemistry. Supported by SPARC (Scholarly Publication and Academic Resources Coalition), *Geochemical Transactions* is intended to provide a low-cost medium for the rapid publication of high-impact articles on all areas of geochemistry. The editor-in-chief, Scott Wood, heads the team of nine, internationally renowned, associate editors.

### Interactive features and web publishing

The poster will show how to make best use of latest webbased technology to enhance your research paper, such as full colour, interactive data (spectra, phase diagrams); crystalstructure visualization, photomicrographs (crossing polars); image enlargement; virtual reality images, mathematical models; movies and animations. For an example see a recent paper by Anderson, which presents truly interactive GC-MS spectral data.<sup>1</sup>

### **Benefits for authors**

A discussion of the frequently asked questions and common concerns or misconceptions about web publication will be presented and the true benefits of publication in *Geochemical Transactions* outlined, including:

- Fast decision from editors (within 2-4 weeks)
- Extremely rapid publication (within 65 days of receipt!)
- Unlimited use of colour and interactive features
- Free pdf files for high-quality reprints
- Wide dissemination of results
- Fully archived articles

#### References

Anderson, K.B., (2001), Geochem. Trans., 2, 21-24